

Año	Mes	Capital	Tasa Interes	meses de mora	Total intereses	Total a pagar	Abono Total	Interes	Abono a capital	Saldo
2016	Ene.	\$ 92.000	2,40	54	\$ 119.232	\$ 211.232		\$ 119.232		\$ 92.000
2016	Feb.	\$ 92.000	2,40	53	\$ 117.024	\$ 209.024		\$ 117.024		\$ 92.000
2016	Mar.	\$ 92.000	2,40	52	\$ 114.816	\$ 206.816		\$ 114.816		\$ 92.000
2016	Mar. Retroac	\$ 21.000	2,40	51	\$ 25.704	\$ 46.704		\$ 25.704		\$ 21.000
2016	Abr	\$ 98.000	2,40	50	\$ 117.600	\$ 215.600		\$ 117.600		\$ 98.000
2016	May	\$ 98.000	2,40	49	\$ 115.248	\$ 213.248		\$ 115.248		\$ 98.000
2016	MayRETROA	\$ 18.000	2,40	48	\$ 20.736	\$ 38.736		\$ 20.736		\$ 18.000
2016	Jun	\$ 98.000	2,40	46	\$ 108.192	\$ 206.192		\$ 108.192		\$ 98.000
2016	Jul	\$ 98.000	2,40	45	\$ 105.840	\$ 203.840		\$ 105.840		\$ 98.000
2016	Ago	\$ 98.000	2,40	44	\$ 103.488	\$ 201.488		\$ 103.488		\$ 98.000
2016	Sep	\$ 98.000	2,40	43	\$ 101.136	\$ 199.136		\$ 101.136		\$ 98.000
2016	Oct	\$ 98.000	2,40	42	\$ 98.784	\$ 196.784		\$ 98.784		\$ 98.000
2016	Nov	\$ 98.000	2,40	41	\$ 96.432	\$ 194.432		\$ 96.432		\$ 98.000
2016	Dic	\$ 98.000	2,40	40	\$ 94.080	\$ 192.080		\$ 94.080		\$ 98.000
TOTALES								\$ 1.338.312		\$ 1.197.000

Año	Mes	Capital	Tasa Interes	meses de mora	Total intereses	Total a pagar	Abono Total	Interes	Abono a capital	Saldo
2017	Ene.	\$ 121.000	2,40	39	\$ 113.256	\$ 234.256		\$ 113.256		\$ 121.000
2017	Feb.	\$ 121.000	2,40	38	\$ 110.352	\$ 231.352		\$ 110.352		\$ 121.000
2017	Mar.	\$ 121.000	2,40	37	\$ 107.448	\$ 228.448		\$ 107.448		\$ 121.000
2017	Abr	\$ 121.000	2,40	36	\$ 104.544	\$ 225.544		\$ 104.544		\$ 121.000
2017	May.	\$ 121.000	2,40	35	\$ 101.640	\$ 222.640		\$ 101.640		\$ 121.000
2017	Jun	\$ 121.000	2,40	34	\$ 98.736	\$ 219.736		\$ 98.736		\$ 121.000
2017	Jul	\$ 121.000	2,40	33	\$ 95.832	\$ 216.832		\$ 95.832		\$ 121.000
2017	Ago	\$ 121.000	2,40	32	\$ 92.928	\$ 213.928		\$ 92.928		\$ 121.000
2017	Sep	\$ 121.000	2,40	31	\$ 90.024	\$ 211.024		\$ 90.024		\$ 121.000
2017	Oct	\$ 121.000	2,40	30	\$ 87.120	\$ 208.120		\$ 87.120		\$ 121.000
2017	Nov	\$ 121.000	2,40	29	\$ 84.216	\$ 205.216		\$ 84.216		\$ 121.000
2017	Dic.	\$ 121.000	2,40	28	\$ 81.312	\$ 202.312		\$ 81.312		\$ 121.000
TOTALES								\$ 1.167.408		\$ 1.452.000

Año	Mes	Capital	Tasa Interes	meses de mora	Total intereses	Total a pagar	Abono Total	Interes	Abono a capital	Saldo
2018	Ene.	\$ 121.000	2,40	27	\$ 78.408	\$ 199.408		\$ 78.408		\$ 121.000
2018	Feb.	\$	2,40	26	\$ 75.504	\$ 196.504		\$ 75.504		\$ 121.000

EXPERIMENTAL RECORD

1. The first experiment was conducted on 10/10/1944. The purpose was to determine the effect of temperature on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction was relatively slow at room temperature.

2. The second experiment was conducted on 10/11/1944. The purpose was to determine the effect of concentration on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction increased with increasing concentration of hydrogen peroxide.

3. The third experiment was conducted on 10/12/1944. The purpose was to determine the effect of catalyst on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction was significantly increased by the addition of a catalyst.

4. The fourth experiment was conducted on 10/13/1944. The purpose was to determine the effect of surface area on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction increased with increasing surface area of the solid reactant.

5. The fifth experiment was conducted on 10/14/1944. The purpose was to determine the effect of pressure on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction was relatively unaffected by changes in pressure.

6. The sixth experiment was conducted on 10/15/1944. The purpose was to determine the effect of pH on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction was relatively unaffected by changes in pH.

7. The seventh experiment was conducted on 10/16/1944. The purpose was to determine the effect of ionic strength on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction was relatively unaffected by changes in ionic strength.

8. The eighth experiment was conducted on 10/17/1944. The purpose was to determine the effect of solvent on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction was relatively unaffected by changes in solvent.

9. The ninth experiment was conducted on 10/18/1944. The purpose was to determine the effect of stirring on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction was significantly increased by stirring.

10. The tenth experiment was conducted on 10/19/1944. The purpose was to determine the effect of light on the rate of reaction between hydrogen peroxide and potassium iodide. The reaction was carried out in a beaker at room temperature (25°C). The rate of reaction was measured by the volume of oxygen gas evolved over a period of 10 minutes. The results showed that the rate of reaction was relatively unaffected by changes in light.

		121.000								
		\$								
2018	Mar.	125.000	2,40	25	\$ 75.000	\$ 200.000		\$ 75.000		\$ 125.000
		\$								
2018	Abr	125.000	2,40	23	\$ 69.000	\$ 194.000		\$ 69.000		\$ 125.000
		\$								
2018	May.	125.000	2,40	22	\$ 66.000	\$ 191.000		\$ 66.000		\$ 125.000
		\$								
2018	Jun	125.000	2,40	21	\$ 63.000	\$ 188.000		\$ 63.000		\$ 125.000
		\$								
2018	Jul	125.000	2,40	20	\$ 60.000	\$ 185.000		\$ 60.000		\$ 125.000
		\$								
2018	Ago	125.000	2,40	19	\$ 57.000	\$ 182.000		\$ 57.000		\$ 125.000
		\$								
2018	Sep	125.000	2,40	18	\$ 54.000	\$ 179.000		\$ 54.000		\$ 125.000
		\$								
2018	Oct	125.000	2,40	17	\$ 51.000	\$ 176.000		\$ 51.000		\$ 125.000
		\$								
2018	Nov	125.000	2,40	16	\$ 48.000	\$ 173.000		\$ 48.000		\$ 125.000
		\$								
2018	Dic.	125.000	2,40	15	\$ 45.000	\$ 170.000		\$ 45.000		\$ 125.000
		\$								\$ 0
								TOTALES	\$ 741.912	\$ 1.492.000

Año	Mes	Capital	Tasa Interes	meses de mora	Total intereses	Total a pagar	Abono Total	Interes	Abono a capital	Saldo
		\$								
2019	Ene.	132.000	2,40	14	\$ 44.352	\$ 176.352		\$ 44.352		\$ 132.000
		\$								
2019	Feb.	132.000	2,40	13	\$ 41.184	\$ 173.184		\$ 41.184		\$ 132.000
		\$								
2019	Mar.	132.000	2,40	12	\$ 38.016	\$ 170.016		\$ 38.016		\$ 132.000
		\$								
2019	Abr	132.000	2,40	11	\$ 34.848	\$ 166.848		\$ 34.848		\$ 132.000
		\$								
2019	Abr EXTR	115.000	2,40	11	\$ 30.360	\$ 145.360		\$ 30.360		\$ 115.000
		\$								
2019	May.	132.000	2,40	10	\$ 31.680	\$ 163.680		\$ 31.680		\$ 132.000
		\$								
2019	Jun	132.000	2,40	9	\$ 28.512	\$ 160.512		\$ 28.512		\$ 132.000
		\$								
2019	Jul	132.000	2,40	8	\$ 25.344	\$ 157.344		\$ 25.344		\$ 132.000
		\$								
2019	Ago	132.000	2,40	7	\$ 22.176	\$ 154.176		\$ 22.176		\$ 132.000
		\$								
2019	Sep	132.000	2,40	6	\$ 19.008	\$ 151.008		\$ 19.008		\$ 132.000
		\$								
2019	Oct	132.000	2,40	5	\$ 15.840	\$ 147.840		\$ 15.840		\$ 132.000
		\$								
2019	Nov	132.000	2,40	4	\$ 12.672	\$ 144.672		\$ 12.672		\$ 132.000
		\$								
2019	Dic.	132.000	2,40	3	\$ 9.504	\$ 141.504		\$ 9.504		\$ 132.000
		\$								\$ 0
								TOTALES	\$ 353.496	\$ 1.699.000

	Intereses de mora	Capital
2016	\$ 1.338.312	\$ 1.197.000
2017	\$ 1.167.408	\$ 1.452.000
2018	\$ 741.912	\$ 1.492.000
2019	\$ 353.496	\$ 1.699.000
Sub total	\$ 3.601.128	\$ 5.840.000
TOTAL	\$ 9.441.128	

1997-1998

the 1990s, the number of people in the world who are under 15 years of age is expected to increase from 1.1 billion to 1.4 billion. The number of people aged 65 and over is expected to increase from 200 million to 400 million. The number of people aged 15 and over is expected to increase from 3.5 billion to 4.5 billion. The number of people aged 15 and over is expected to increase from 3.5 billion to 4.5 billion. The number of people aged 15 and over is expected to increase from 3.5 billion to 4.5 billion.

the 1990s, the number of people in the world who are illiterate has increased from 1.2 billion to 1.5 billion. The number of illiterate people in the world is expected to reach 1.7 billion by the year 2015. The number of illiterate people in the world is expected to reach 1.7 billion by the year 2015.

Figure 1. The effect of the concentration of the solution on the adsorption capacity of the polymer sorbent. The conditions were as follows: temperature, 20°C; pH, 7.0; time, 6 h; amount of sorbent, 0.1 g; volume of solution, 10 mL; initial concentration of Cu(II), 1 mg/L.

6. *Chlorophyll a* and *Chlorophyll b* contents were determined by the method of Arar and Cook (1987).

[illegible]